

JOINING STEELS BY MEANS OF AN AMORPHOUS WELDING PROCESS

B. ARCONDO⁺, F. AUDEBERT, C. EPELBAUM, M. FONTANA and H. SIRKIN

Laboratorio de Sólidos Amorfos, Facultad de Ingeniería, UBA. Paseo Colón 850 (1063) Buenos Aires, Argentina

⁺ E-mail address: barcond@fi.uba.ar

Abstract— The *Laboratorio de Sólidos Amorfos* started research on amorphous bonding processes in 1998. Since the beginning of 2000 a three year project devoted to this subject “Joining metals by amorphous metals diffusion” has been subsidized by the ANPCyT[‡] and Siderca. In this work the antecedents of the project are summarized, the welding method is described and the results obtained up to now are reported.

Keywords— amorphous bonding, welding, amorphous metals, steel.

I. INTRODUCTION

The *Laboratorio de Sólidos Amorfos (LSA)* was born as a consequence of the basic research on liquid semiconductors performed in the 70's in the Physics Department. In the early 80's a group devoted to the study of semiconducting glasses and amorphous metals produced by means of rapid solidification techniques (Quintana et al. 1979), and analyzed employing X-ray diffractometry and Mössbauer spectrometry consolidated as a new laboratory, the LSA. The early works were devoted to the glass forming ability of several binary and ternary chalcogenide based systems (Quintana et al. 1982, Arcondo et al. 1984, Arcondo et al. 1985 a-b, Fontana et al. 1992). By the end of this period, the glass forming mechanisms of a collection of Mg based alloy were studied. These alloys are MgSnX (X= Cu, Ni, Zn and Ga) and MgZnPb (Mingolo et al. 1986, Sirkin et al. 1987, Mingolo et al. 1989, Vicente et al. 1991, Arcondo et al. 1991a, Vicente et al. 1992, Fontana and Arcondo, 1994, Fontana and Arcondo, 1995, So-moza et al. 1995).

Once the rapid solidification melt spinning technique was implemented in the *Laboratorio de Sólidos Amorfos*, work on Fe or Ni based multicomponent systems began. Among them FeB, NiB, FeNiB, FeBSn, NiBSn, FeNiBSn (Boudard et al. 1991, Arcondo et al. 1991b, Arcondo et al. 1994 a-d)

After this, a new stage started, characterized by the diversification of the properties studied and the opening to the collaboration with European research groups. Two main lines, metallic and chalcogenide glasses, have

been developed. The amorphous metals line, included as well the study of nanostructured alloys obtained by means of a controlled crystallization process of FeSiB (Moya et al. 1997a) and several light amorphous metals (Audebert et al. 1997a). The glass forming ability, the structure (Audebert et al. 1997b), the mechanical (Moya et al. 1997b) (Young's modulus, microhardness, tensile strength, plasticity and fracture) and magnetic (coercivity, saturation, remanence and permeability) (Moya et al. 1999) properties, the corrosion resistance (Audebert et al. 1998), in acid and basic media, and the crystallization processes have been analyzed. The systems studied were FeSiBSn, FeSiBNb, FeSiBNbSn, FeSiBNbCu, FeSiBNbCuX (X= Al, Cr, Ag, C, etc.), Al-FeX (X=Nb, Sb, MM: Mishmetal, a Rare Earths mixture) and Mg(Cu, Al)MM. The chalcogenide glasses line main interests have been the structure and electric and mass transport studies and the crystallization kinetics (Fontana et al. 2000a, Fontana et al. 2000b).

In the early 90's a new project tending to the development of a centrifugal atomization device to produce metallic powders with the aim to obtain sintered pieces of high mechanical resistance was initiated. The systems studied were CoCrWC and FeMnAlSiC (Ozols et al. 1999, Ozols et al. 1996). This line is oriented nowadays towards the elaboration of soft magnetic nucleus employing preisolated powder particles. The systems under study are NiFeMo, FeCoV and NiFeCo.

The results of these works are reported in more than 100 papers in international journals and the contribution of the *Laboratorio de Sólidos Amorfos* to the formation of human resources has been expressed in 10 doctoral thesis in Physics and Engineering.

The above mentioned lines are supplemented at the present time with a technological project oriented to the development of metallic joints employing amorphous metals layers and a recently initiated project oriented to the modification of metallic or semiconducting surfaces employing laser discharges.

II. METHODS

A. Amorphous Welding Method

Let us assume that two bars of a metal α are aligned with their slightly polished butts in contact with a thin amorphous layer L (Fig. 1). The composition of L is not far from an eutectic point which components are α and β (Fig. 2). The melting temperature of L (T_L) is lower

[‡] Agencia Nacional para Promoción de la Ciencia y la Tecnología.

than the melting temperature of α (T_α). If the temperature in the joining zone is raised to T_j at the higher attainable rate, being $T_L < T_j < T_\alpha$, a transient liquid phase gap, with an initial composition x_L , is originated. As the temperature T_j is maintained constant the following process will take place: i) the dissolution of the α atoms from the bars into the liquid gap takes place, widening the gap ii) when the liquid composition x_M is attained, the solidification starts and the thinning of the gap proceeds up to its total disappearance, iii) the completion of the solid state diffusion is performed (Cain et al. 1997, Tuah-Poku et al. 1988) (Fig. 3). Moreover, if the diffusion coefficient of β in α is large, the picture must include the simultaneous diffusion of β into the bars with the consequent shortening of the process.

In the case that α bars are single crystalline, a planar diffusion front can be modeled but being the bars polycrystalline the diffusion along grain boundaries has to be analyzed (Kaur et al. 1995). The amorphous nature of the layer is not a precondition for the transient liquid phase bonding process but, due to its large toughness (when compared with the corresponding crystalline state), small thickness amorphous layers are easy to handle and the contact between the layer and the bars is improved, particularly when pressure is applied.

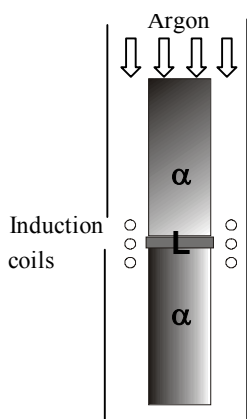


Figure 1. Schematic welding device. Two bars of metal α and an amorphous interlayer L placed in an induction furnace.

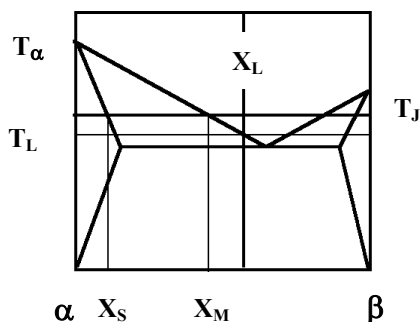


Figure 2. Phases equilibrium diagram of the α - β system. T_j is the isothermal welding temperature, X_L the initial composition of the transient liquid gap, X_M the compo-

sition of the liquid gap as a solid phase of composition X_S precipitates.

B. Experimental Procedure

Fe rich Fe-B and Fe-Si-B amorphous metals have been employed as interlayers in transient liquid phase bonding processes performed to join two Carbon Manganese steel bars α^* with Fe the majority component α . The bars were aligned with the polished butts of both pieces in contact with the amorphous layer and located into the coil of four turns in an induction furnace under Ar flux. The heated area is a fringe of 10 mm to both sides of the line of joint. The temperature has been raised at the highest attainable rate to $T \approx 1300^\circ\text{C}$ and then maintained constant. The welding processes modelling must include a) layer thickness of about $25\mu\text{m}$ with melting temperatures $T \approx 1100^\circ\text{C}$, b) the dissolution of the Fe, and other species atoms, from the bars into the liquid gap while the B atoms migrate from the liquid towards the bars with a grain boundaries diffusion coefficient much higher than the bulk diffusion constant, despite the grain boundary segregation effects on B diffusivity.

III. RESULTS

Bonds performed employing Fe-B amorphous ribbons are depicted in Figs. 3-5. Figure 3 shows an optical micrograph of the welding zone. The vertical black line represents the thickness ($25\mu\text{m}$) of the amorphous Fe-B interlayer. The remaining welding traces denote an incomplete diffusion as well as the role of the grain boundaries in the process. The microstructure of the welding zone is a consequence of both the boron diffusion and the thermal process.

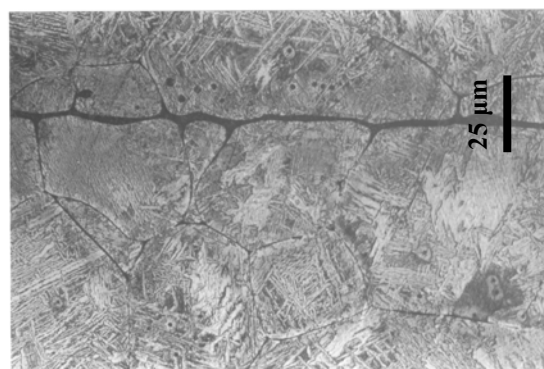


Figure 3. Optical micrograph of the welding zone. The thickness ($25\mu\text{m}$) of the eutectic Fe-B amorphous ribbon is marked with a black vertical line.

Figures 4 and 5 correspond to scanning electron micrographs of Carbon Manganese steel bars welded employing eutectic Fe-B amorphous ribbons with a $25\mu\text{m}$ thickness. The thermal process does not differ significantly from the above mentioned. Boron has been detected by EDAX in all the positions marked as A, B and C denoting inter grain as well as intra grain diffusion.

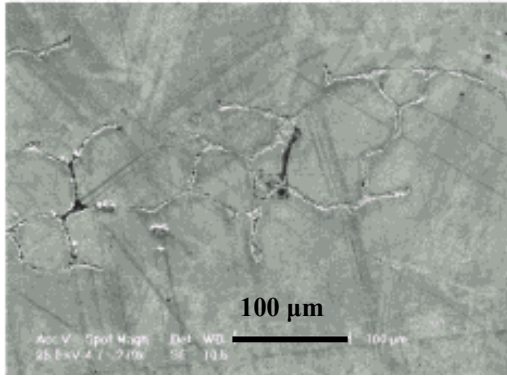


Figure 4. Scanning electron micrograph of the welding zone of two steel bars joined employing eutectic Fe-B amorphous ribbons (25μm thickness) under Ar flux.

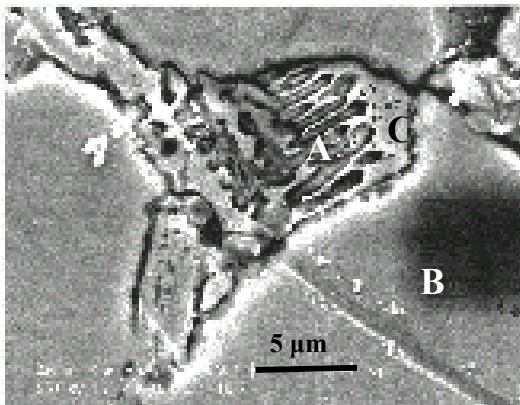


Figure 5. Scanning electron micrograph of the welding zone (grain border) of two C-Mn steel bars joined employing eutectic Fe-B amorphous ribbons (25μm thickness) under argon flux.

Average Vickers microhardness has been measured applying 200g loads during 15 seconds in several points in the whole width of the bars. A significant increasing of the Vickers microhardness has been observed in the joint zone, as compared with that of the C-Mn steel bars (about 260 Hv) (Figure 6a). This fact is a consequence of the formation of martensite and, probably, borides. The first one is formed by means of a cooling in air due to the solid solubility of boron. Borides, although their presence could not be confirmed by X Ray diffraction, are not discarded in the grain boundaries as a consequence of the high boron content of the amorphous ribbons (17 at. % B). An incomplete diffusion of boron due to the presence of borides in the grain boundaries may induce the snappy variations of the microhardness observed along the affected zone.

Fe-Si-B amorphous ribbons (25μm thick and 5mm wide) have been obtained with the aim to improve the mechanical properties of the joints. In fact, Vickers microhardness of the welding zone decreases as can be seen in Figure 6b. Also, smooth variations of the microhardness can be observed denoting a more complete diffusion as the boron content is significantly lower.

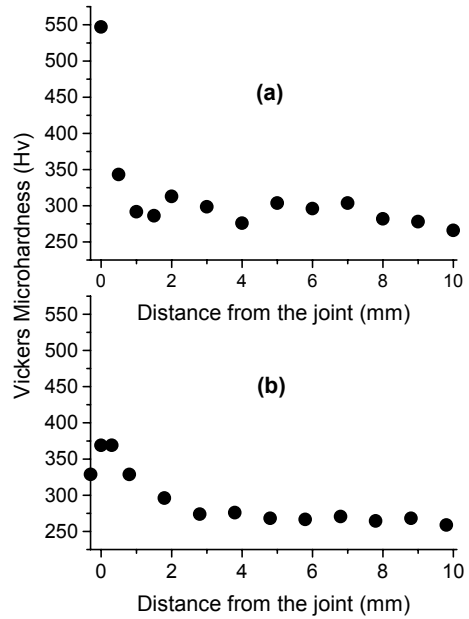


Figure 6. Average values of Vickers microhardness measured along the welded bars. On top, measurements performed on a sample welded employing Fe-B amorphous ribbons, under these, measurements performed on a sample welded with Fe-Si-B interlayer.

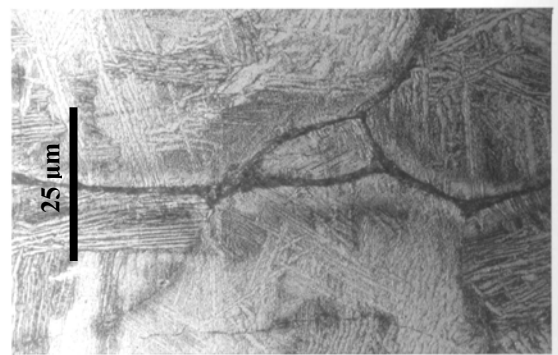


Figure 7. Optical micrograph of the welding trace of two C-Mn steel bars joined employing an amorphous Fe-Si-B ribbon (25μm thickness) under Ar flux. The black line corresponds to the original thickness of the amorphous interlayer.

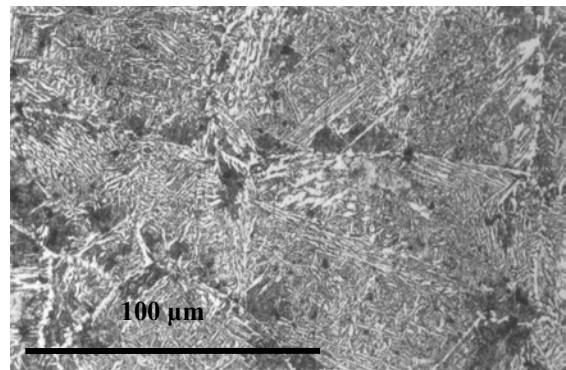


Figure 8. Optical micrograph of the heat affected zone in the central region of the bar, 3 mm below the welding trace.

After the welding process, the samples were annealed to transform the martensite and to complete the solid state diffusion.

Figures 7 and 8 depict the welding zone and the thermally affected region, 3mm above the joint, of samples welded employing Fe-Si-B amorphous ribbons.

The extension of the thermally affected zone depends on several variables, the temperature of the process, its duration and the height of the heating coils. In the case of the sample depicted in Fig. 7 and 8, four coils were employed.

IV. CONCLUSIONS

In view of the results presented here, this TLP bonding method employing amorphous interlayers is a suitable method for welding steel pieces of different geometries i.e. bars or pipes.

The duration of the process is in the order of a few minutes, the temperature is not much higher than the melting point of the interlayer. No high pressures are involved. Nevertheless the significant parameters must be optimized in order to improve the mechanical properties of the joint and also its corrosion resistance. An aspect that has not been explored yet in this project

ACKNOWLEDGEMENTS

The authors acknowledge Mr. A. Carnero, Mr. E. Carretero and Eng. E. Castro their collaboration in the technical aspects of this work.

Also, the financial support of the Universidad de Buenos Aires, the Agencia Nacional para la Promoción de la Ciencia y la Tecnología and Siderca, through the PICTO 4603, is greatly acknowledged.

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Received May 15, 2001.

Accepted for publication: February 15, 2002.

Recommended by Subject Editor E. Dvorkin.